

Reinterpretation of the conjectured Jahn-Teller switch of MnF_6^{3-} complexes in Na_3MnF_6 under pressure

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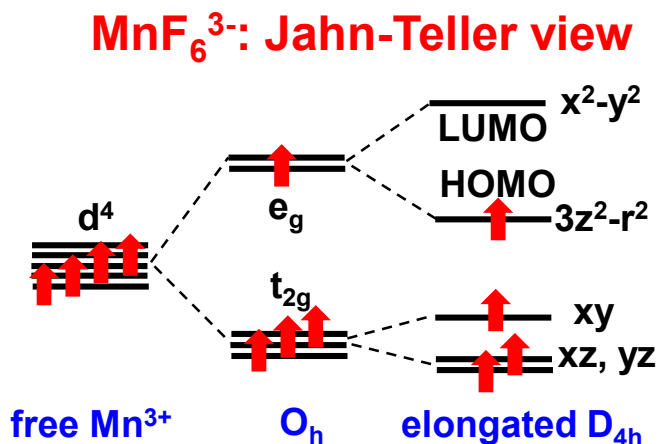
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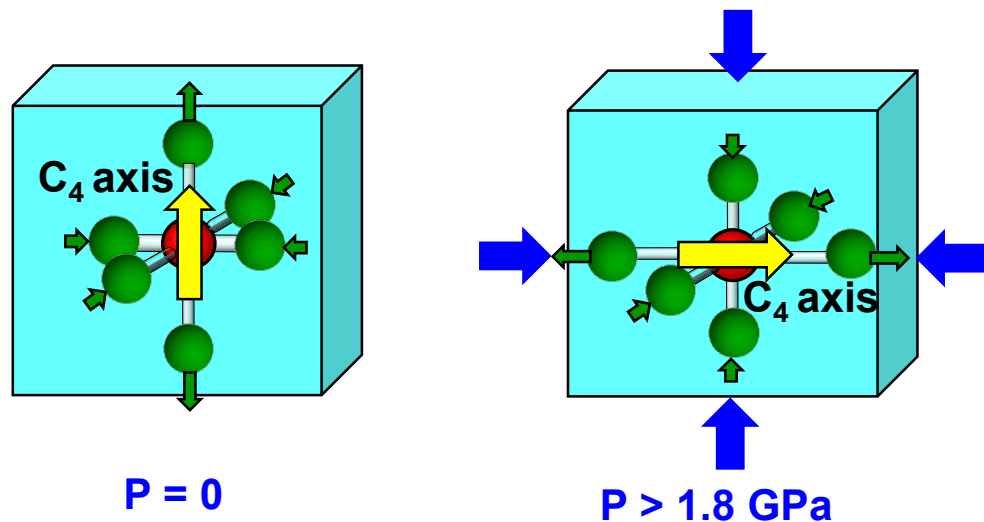
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Jahn-Teller switch under pressure

- Many works invoke the existence of a **Jahn-Teller switch under hydrostatic pressure** in solids containing complexes of transition metal cations as Cu^{2+} (d^9 configuration) or Mn^{3+} (d^4)
- Examples:** Na_3MnF_6 , $(\text{NH}_4)_2\text{Cu}(\text{H}_2\text{O})_6(\text{SO}_4)_2$, $\text{Cs}_2\text{Zn}(\text{ZrF}_6)_2 \cdot 6\text{H}_2\text{O}:\text{Cu}^{2+}$, $\text{CuF}_2(\text{H}_2\text{O})_2(\text{pyz})$ (pyz = pyrazine), CuWO_4 , etc.
- Conjetured idea:** $\text{Na}_3\text{MnF}_6 \rightarrow \text{MnF}_6^{3-}$ complexes (d^4 , $S = 2$) tetragonally elongated (D_{4h}) by **Jahn-Teller effect**

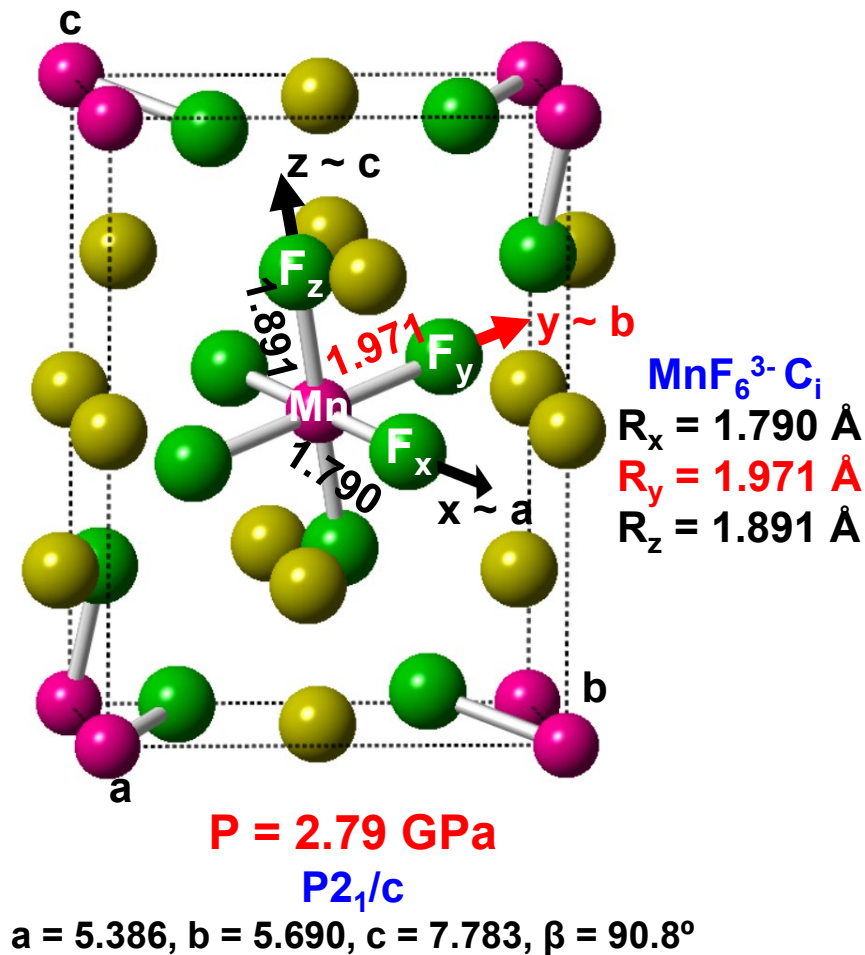
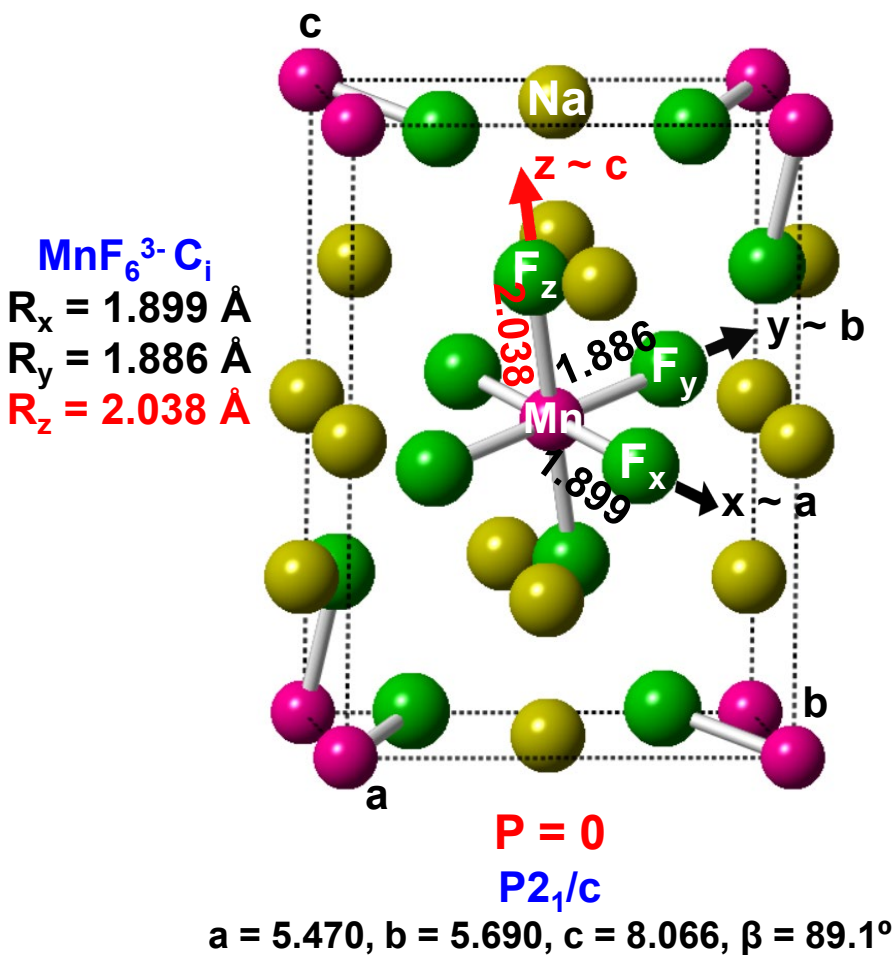


MnF_6^{3-} : switch of the C_4 axis under P



Na₃MnF₆ monoclinic P2₁/n (standard P2₁/c)

- ❑ **Experimental data:** Carlsson et al., Inorg. Chem. 37, 1486 (1988)
- ❑ **MnF₆³⁻ complexes:** no D_{4h} elongated but **triclinic C_i**, no C₄ axis
- ❑ **Switch of the long Mn-F bond under P = 2.79 GPa: z → y**



Goals and Tools

□ 3 goals

1. To show that **there is not a Jahn-Teller effect** in MnF_6^{3-} complexes of Na_3MnF_6 (the same for other non-cubic crystals)
2. To show that the long axis switch is due to the **anisotropy** of the crystal
3. To **assign the 3 observed peaks in the optical d-d spectrum**

□ Basic tools

- **First-principles DFT periodic calculations:** Crystal code, hybrid XC functionals
- **First-principles DFT calculations on a MnF_6^{3-} complex with embedding:** ADF code, hybrid XC functionals
- **Symmetry**

Na₃MnF₆: calculated geometry

❑ **Optimized geometries** for $P = 0, 2.79$ GPa

❑ Lattice parameters (a, b, c) and Mn-F distances in Å, β angle in degrees

		a	b	c	β	R_x	R_y	R_z
P = 0	Experim.	5.471	5.683	8.073	88.9	1.862	1.897	2.018
	Calculated	5.460	5.646	8.137	88.5	1.864	1.880	2.069
P = 2.79	Experim.	5.386	5.690	7.783	89.3	1.790	1.971	1.891
	Calculated	5.381	5.720	7.713	88.4	1.861	2.041	1.866

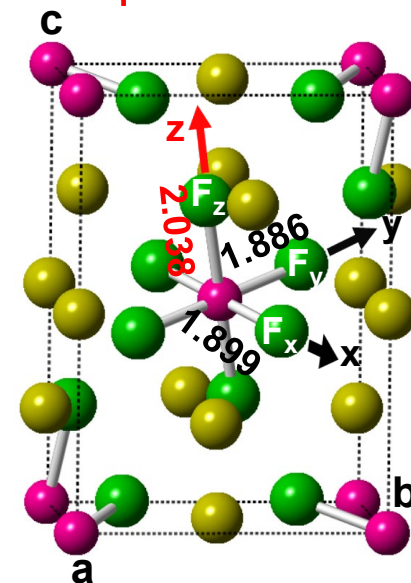
lattice variation

-0.10 **+0.01** **-0.29 Å**
 $R_x \downarrow$ $R_y \uparrow$ $R_z \downarrow$

very anisotropic

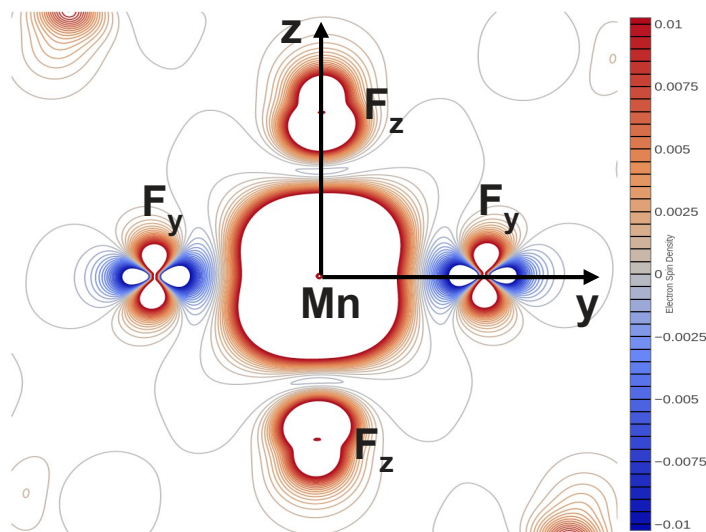
❑ Good agreement with experimental geometries

❑ Calculations reproduce the switch under P



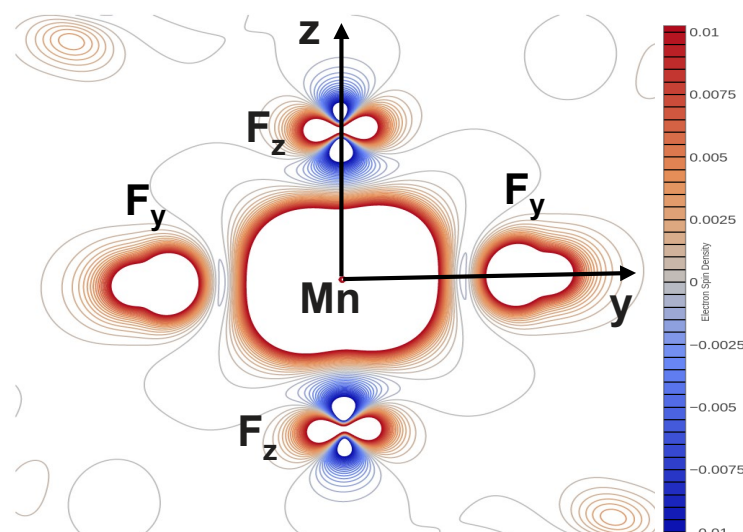
Na₃MnF₆: calculated spin density

□ Calculated spin density (up - down) of a MnF₆³⁻ complex



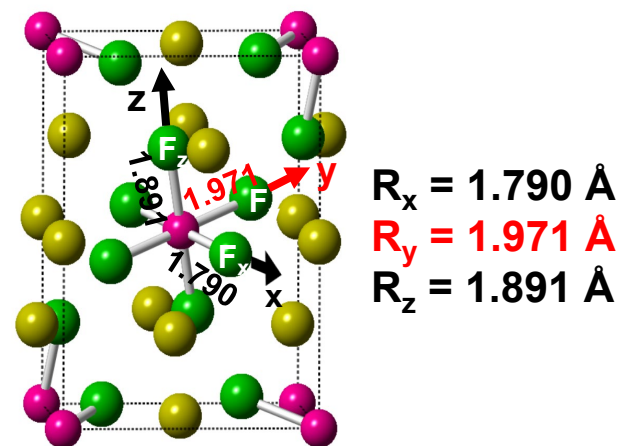
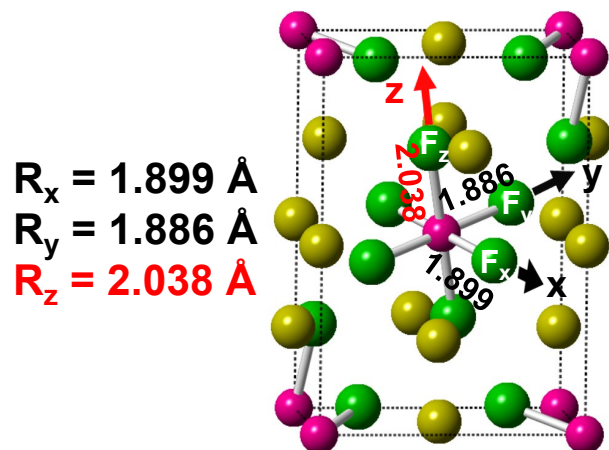
P = 0

HOMO: dominant $3z^2-r^2$ character



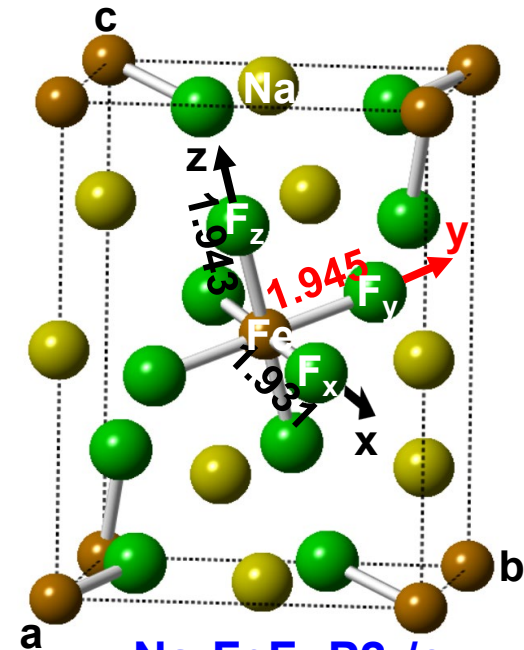
P = 2.79 GPa

HOMO: dominant $3y^2-r^2$ character



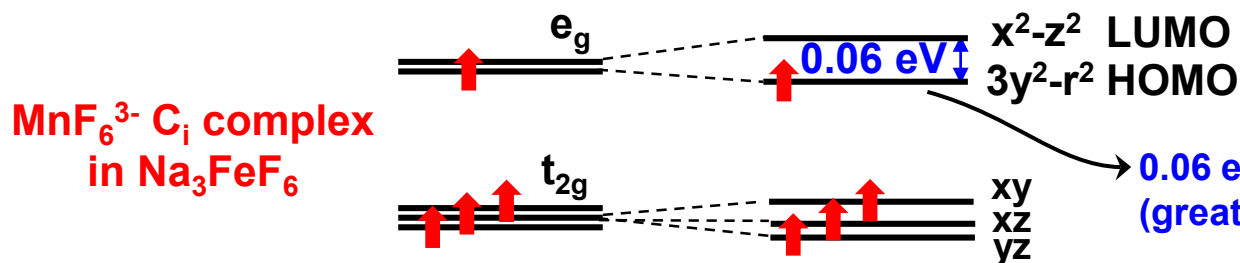
Na₃MnF₆: no Jahn-Teller effect

- ❑ Jahn-Teller effect is very restrictive: it requires a high symmetry **parent phase where MnF₆³⁻ complexes had orbital degeneration**
- ❑ **Parent phase of Na₃MnF₆: killing any vibronic coupling**
- ❑ **Substitute all Mn³⁺ (d⁴ open shell) → Fe³⁺ (d⁵, semiclosed shell), same ionic radii, 0.785 Å**
- ❑ **Geometry optimization of Na₃FeF₆ fixing P2₁/n group** ⇒ equal or greater symmetry (no lower)
- ❑ **Na₃FeF₆ parent phase no Jahn-Teller effect**
 - Same P2₁/n symmetry of Na₃MnF₆
 - FeF₆³⁻ complexes: same triclinic C_i symmetry
 - Complexes elongated along y
 - Gap (x²-z²) - (3y²-r²) of 0.06 eV: no degeneration



Na₃FeF₆ P2₁/c

R_x = 1.931 Å
 R_y = 1.945 Å
 R_z = 1.943 Å

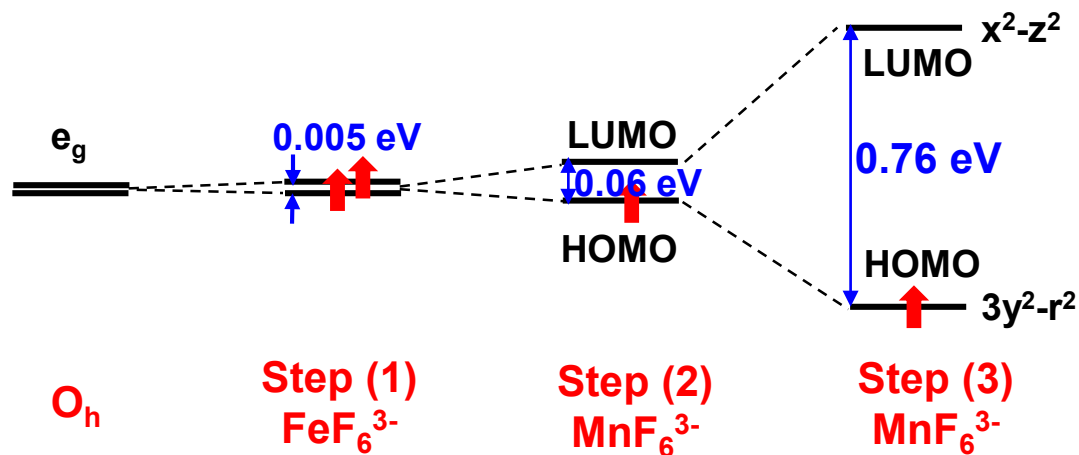


0.06 eV is not quasi-degeneration
 (greater than Jahn-Teller barrier)

One MnF_6^{3-} complex in Na_3FeF_6

step	system	calculation	Complex	R_x (Å)	R_y (Å)	R_z (Å)	Δ (eV)
(1)	Na_3FeF_6	full optimization	FeF_6^{3-}	1.931	1.945	1.943	0.005
(2)	$\text{Na}_3\text{FeF}_6:\text{Mn}^{3+}$	Na_3FeF_6 param. and distances	MnF_6^{3-}	1.931	1.945	1.943	0.06
(3)	$\text{Na}_3\text{FeF}_6:\text{Mn}^{3+}$	Na_3FeF_6 param., optimize distances	MnF_6^{3-}	1.856	2.045	1.895	0.76
(4)	$\text{Na}_3\text{FeF}_6:\text{Mn}^{3+}$	increase c 1.1% optimize distances	MnF_6^{3-}	1.866	1.892	2.049	

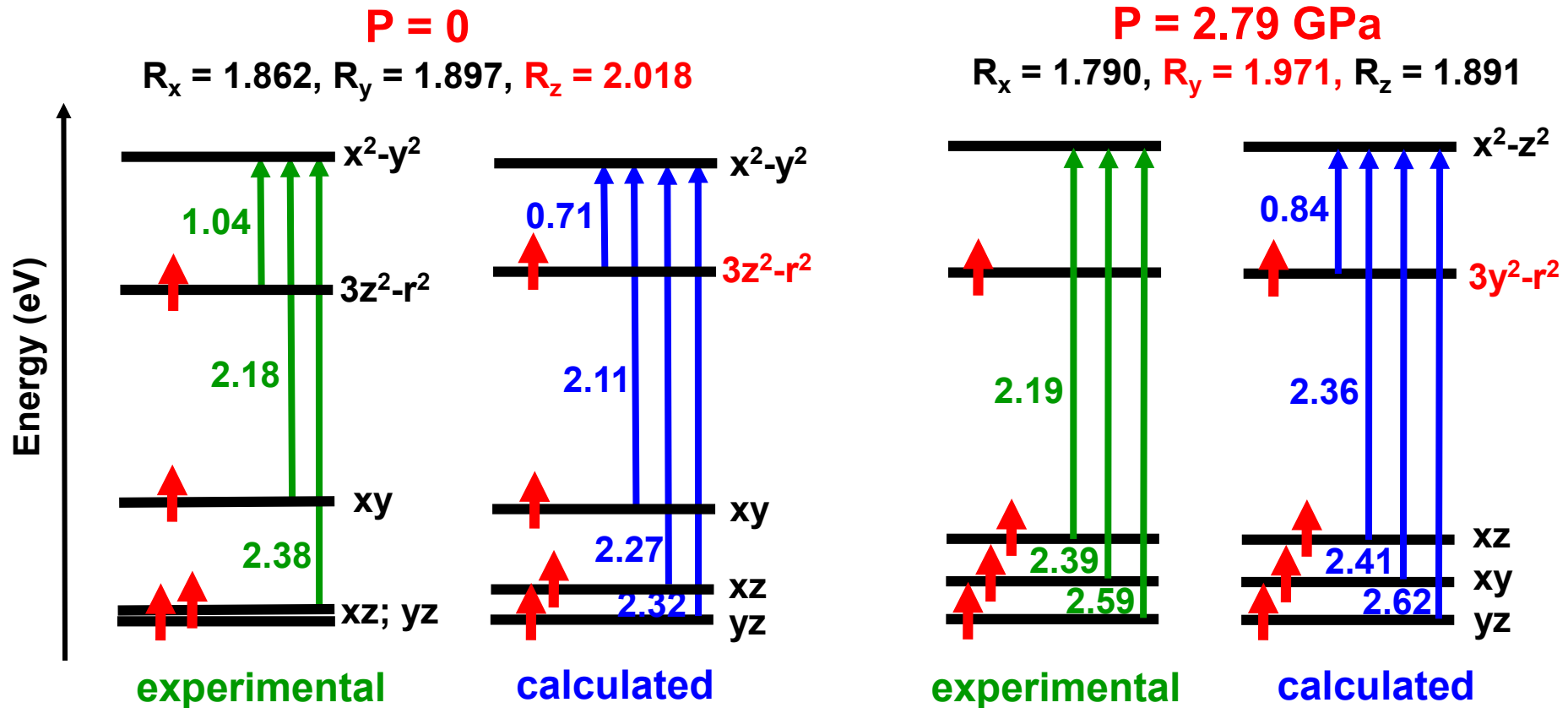
splitting Δ of e_g in a C_i complex in Na_3FeF_6



Pseudo-Jahn-Teller effect
coupling HOMO-LUMO
through a_g modes

MnF₆³⁻: d-d transitions, P = 0, 2.79 GPa

- Calculated energies (in eV) for a C_i MnF₆³⁻ complex in Na₃MnF₆ at P = 0, 2.79 GPa
- Experimental data: Carlsson et al., Inorg. Chem. 37, 1486 (1988)



More information

- ❑ I. Sánchez-Movellán, D. Carrasco-Busturia, J. M. García-Lastra, J. A. Aramburu, P. Garcia-Fernández, M. Moreno, *Chem. Eur. J.* 28, e202200948 (2022)